

# STRONTIUM

By Joyce A. Ober

Prompted by the expected demand for additional strontium carbonate for the television glass industry, new strontium carbonate production facilities were completed in the early 1990's and celestite production increased, but not as dramatically as the earlier increases in strontium carbonate capacity. Significant increases in world production of celestite indicate that the capacity utilization of strontium carbonate facilities is beginning to expand in response to the additional demand for strontium carbonate required to supply newly completed television glass plants. Chemical Products Corp. (CPC) of Cartersville, GA, is the only U.S. producer of strontium compounds from celestite. CPC produces strontium carbonate from imported Mexican ore; no celestite mines are active in the United States.

Strontium occurs commonly in nature, averaging 0.034% of all igneous rock; however, only two minerals, celestite (strontium sulfate) and strontianite (strontium carbonate), contain strontium in sufficient quantities to make its recovery practical. Of the two, celestite occurs much more frequently in sedimentary deposits of sufficient size to make development of mining facilities currently attractive. Strontianite would be the more useful of the two common minerals because strontium is used most commonly in the carbonate form, but few deposits have been discovered that are suitable for development. Only one strontianite mine is believed to operate in the world.

## Legislation and Government Programs

Government stockpiling of celestite began in 1942 to provide a secure supply of celestite for the production of strontium compounds required for defense applications during World War II. Celestite purchase specifications issued in 1960 for the U.S. National Defense Stockpile established quality requirements at greater than 95% strontium sulfate content with less than 1.5% calcium sulfate and less than 2% barium sulfate.

In 1963, the celestite stockpile was determined to be unnecessary, and the General Services Administration began selling stockpile material. All the stockpile-grade celestite was sold by 1973, with the remaining material grading less than 91% strontium sulfate and more than 4% calcium sulfate and some with more than 10% barium sulfate.

The stockpile currently contains approximately 12,000 tons of celestite, all of which have been authorized for disposal. The low quality of the material remaining in the stockpile makes it undesirable as a raw material for strontium carbonate production. Reports issued by the Defense Logistics Agency, Defense National Stockpile Center—the agency now responsible for managing stockpile sales—list the celestite as

valueless.

## Production

Although there have been no active celestite mines in the United States since 1959, celestite deposits have been identified nationwide. During World War II, domestic mining of celestite resources was conducted in California and Texas. U.S. celestite mines had at that time been inactive since World War I, and strontium minerals were imported to satisfy domestic demand.

The sole U.S. strontium carbonate producer voluntarily provided domestic production and sales data to the U.S. Geological Survey (USGS). Production and stock data, however, were withheld from publication to avoid disclosing company proprietary data. (*See table 1.*) CPC is the only domestic company that produces strontium carbonate from celestite; the company also produces strontium nitrate. All the celestite CPC used in 1995 was imported from Mexico. CPC owns and operates a second strontium carbonate plant in Reynosa, Tamaulipas, Mexico. The company uses the black ash method of strontium carbonate production at both of its facilities.

The black ash method and the soda ash method are the two most common recovery techniques. The black ash method, known alternatively as the calcining method, produces chemical-grade strontium carbonate, which contains at least 98% strontium carbonate. The soda ash or direct conversion method produces technical-grade strontium carbonate, containing at least 97% strontium carbonate. The first step in the black ash process involves mixing the crushed and screened celestite with powdered coal. The mixture is then heated to about 1,100° C, expelling oxygen in the form of carbon dioxide from the insoluble strontium sulfate to form water-soluble strontium sulfide. Strontium sulfide is dissolved in water and the resulting solution filtered. Carbon dioxide is then passed through the solution or soda ash is added, forming and precipitating strontium carbonate from solution. The precipitated strontium carbonate is then filtered, dried, ground, and packaged. The byproduct sulfur from the process is recovered as elemental sulfur or as other byproduct sulfur compounds.

In the soda ash method, ground celestite is washed and most of the water removed. The thickened mixture is combined with soda ash and treated with steam for 1 to 3 hours. During this time, celestite and soda ash react to form strontium carbonate and sodium sulfate. Sodium sulfate is water soluble, making it possible to separate the insoluble strontium carbonate by centrifuging. The black ash method is the preferred means of strontium carbonate production due to the higher grade product,

and most new production facilities employ black ash technology.

Several U.S. companies produced strontium metal and compounds from strontium carbonate. Calstron Corp. of Memphis, TN, a subsidiary of KB Alloys Inc. of Reading, PA, produced strontium metal. Mallinkrodt Inc. of St. Louis, MO, produced strontium chloride, and Mineral Pigments Corp. of Beltsville, MD, produced strontium chromate. A few other companies produced downstream strontium compounds, but on a limited scale.

## Consumption

The USGS estimated the distribution of strontium compounds by end use. Of the 11 operations to which a survey request was sent, all responded. The information collected from this survey and the information provided by the U.S. Bureau of the Census on strontium trade were the basis for the end-use estimates shown in table 2.

In 1996, more than 80% of all strontium was consumed in ceramics and glass manufacture, primarily in television faceplate glass and ceramic ferrite magnets, and in smaller amounts in other ceramic and glass applications. Over the past 20 years, production of faceplate glass for color television picture tubes has become the major consumer of strontium.

All color televisions and other devices containing color cathode-ray tubes sold in the United States are required by law to contain strontium in the faceplate glass of the picture tube to block X-ray emissions. Major manufacturers of television picture tube glass incorporate about 8%, by weight, strontium oxide in their glass faceplate material. Added to the glass melt in the form of strontium carbonate, it is converted to strontium oxide. In addition to blocking X-rays, the strontium improves the appearance of the glass, increasing the brilliance and improving the quality of the picture.

Permanent ceramic magnets are another large end use for strontium compounds, in the form of strontium ferrite. These magnets are used extensively in small direct current motors used in automobile windshield wipers, loudspeakers, other electronic equipment, toys, and magnetically attached decorative items. Strontium ferrite magnets have high coercive force, high thermal and electrical resistivity, and are chemically inert. They retain their magnetism well, are not adversely affected by electrical currents or high temperatures, do not react with most chemical solvents, resist demagnetization, and are low density.

One of the most consistent and continuing applications for strontium has been in pyrotechnic devices. Strontium burns with a brilliant red flame, and no other material has been found to be better in this application.

The strontium compound used most frequently in pyrotechnic devices was strontium nitrate. Strontium carbonate, strontium oxalate, strontium sulfate, and strontium chlorate can be used in pyrotechnic applications, but strontium nitrate is used in significantly larger quantities than any of these. Pyrotechnic devices are used in military and nonmilitary applications. Military pyrotechnic applications that contained strontium included tracer ammunition, military flares, and marine distress

signals. Nonmilitary applications include warning devices and fireworks.

Strontium is used to remove lead impurities during the electrolytic production of zinc. The addition of strontium carbonate in sulfuric acid to the electrolyte reduces the lead content of the electrolyte and of the zinc that is deposited on the cathode.

Strontium chromate is an additive to corrosion-resistant paint. It is an effective coating for aluminum, most notably on aircraft fuselages and ships. These paints are used to some degree on aluminum packaging to prevent package corrosion.

Strontium metal is a very limited part of total strontium consumption. Small amounts of strontium are added to molten aluminum to improve the castability of the metal, making it more suitable for casting items that have been traditionally made from steel. The addition of strontium to the melt improves the machinability of the casting. The use of cast aluminum parts is currently gaining popularity in the automotive industry because the weight reduction of the automobile from using cast aluminum parts instead of steel improves gas mileage.

Other end uses consume only small amounts of strontium and strontium compounds. As mentioned previously, the presence of strontium in glass applications improves the brilliance of the glass. It also improves the quality of certain ceramic glazes as well as eliminates the toxicity that may be present in glazes containing lead or barium. Strontium titanate is sometimes used as substrate material for semiconductors and in some optical and piezoelectric applications. Strontium chloride is used in toothpaste for temperature-sensitive teeth. For this application, impurities must be strictly controlled, with limits for some of them in the parts per million range. Strontium phosphate is used in the manufacture of fluorescent lights, and the entire range of strontium chemicals was used in analytical chemistry laboratories. (*See table 2.*)

## Prices

The average customs value for celestite imported from Mexico was about \$67 per ton, 6% lower than the average value in 1995. The average unit customs value of imported strontium carbonate was \$0.61 per kilogram, the same as that in 1995. The corresponding value for strontium nitrate was \$1.98 per kilogram, an increase of more than 100%.

## Foreign Trade

According to reports from the U.S. Bureau of the Census, exports of strontium compounds decreased about 10% from the levels reported in 1995. Imports of celestite—all from Mexico—were 26,400 tons, a decrease of about 9% from the 1995 level. Although celestite import figures vary significantly from year to year, this appears to be attributable to a 2-year cycle of celestite imports; however, during the past 2 years, trade data indicate a developing trend toward lower celestite imports. Over the past 5 years, annual celestite imports averaged about 33,000 tons (14,000 tons contained strontium)

per year.

Mexico continued as the most important source for imported strontium compounds; Germany was second. Imports of strontium carbonate in 1996 were about the same as those in 1995, with imports from Mexico comprising 89% of total carbonate imports. Imports of strontium nitrate were 20% lower than those in 1995; strontium nitrate imports are extremely small when compared with imports of strontium carbonate. (*See tables 3 and 4.*)

## World Review

In most instances, celestite deposits occur in remote, undeveloped locations far from population centers in areas where inexpensive labor is available for mining. Huge deposits of high-grade celestite have been discovered throughout the world. Strontium commonly occurs along with barium and calcium, two elements with chemical properties very similar to strontium, thus making separation difficult. Because removing many impurities from celestite is difficult and energy intensive, current strontium chemical producers require material to contain at least 90% strontium sulfate. Most of the currently operating celestite facilities can produce sufficient supplies with only minimal processing necessary to achieve acceptable specifications. Hand sorting and some washing are all that are necessary at many strontium mines; a few operations use gravity separation, froth flotation, or other methods to beneficiate ore.

Leading celestite producers are Mexico, Spain, Turkey, and Iran, in decreasing order of importance. Significant quantities of celestite are produced in China and the former U.S.S.R.; however, not enough information is available to make any estimate of location, number, or size of mines. Celestite is produced in smaller quantities in Algeria, Argentina, and Pakistan.

Detailed information on most world resources is not readily available because very little information on exploration results has been published. Other deposits may be well identified but are in countries from which specific mineral information is not easily obtained.

Until World War II, practically all celestite produced worldwide was from the United Kingdom, and for several years following the end of the war, the United Kingdom maintained its importance. Depleted reserves, encroaching civilization, and the discovery of vast deposits in other countries resulted in the cessation of production in the United Kingdom. Any reported United Kingdom celestite production since 1992 was due to the reprocessing of mine tailings from the exhausted mine. (*See table 5.*)

Production facilities for strontium compounds and/or metal are located in Canada, China, Germany, Japan, the Republic of Korea, Mexico, Poland, the United States, and the former U.S.S.R.

## Outlook

Sales of televisions and computer monitors in the United States continued at a high rate. Faceplate glass plants continued to operate at capacity and imports of glass continued. Because U.S. television glass producers are operating at capacity, domestic consumption of strontium carbonate for this end use will remain relatively level until additional glass capacity is installed. However, worldwide demand for strontium carbonate will increase to meet the demand for U.S. glass imports.

Growth in television sales is expected to continue and larger screens are expected to increase in popularity, and so demand for strontium carbonate for television application should expand. Growth in other markets will probably continue at their current, slower rate.

Development of a technology to produce an affordable flat television display could severely reduce the demand for strontium carbonate, but this is not expected in the near future. Although a high-quality, large, flat screen is not yet available, small models are, and research continues to seek improvements in the technology. The question remains whether a new display system can be developed that will be economically attractive to the general public. Initial devices are expected to find application in military hardware and other sophisticated medical and scientific instrumentation. Flat screen display systems may eventually replace cathode-ray tubes, and, at that point, strontium producers may experience a serious setback.

## SOURCES OF INFORMATION

### U.S. Geological Survey Publications

Strontium. Ch. in *Mineral Commodity Summaries*, annual.

Strontium. Ch. in *Minerals Yearbook*, annual.<sup>1</sup>

### Other

American Ceramic Society Bulletin, monthly.

Chemical Market Reporter, weekly.

Engineering and Mining Journal, monthly.

Industrial Minerals (London), monthly.

Mining Annual Review (London).

Mining Engineering, monthly.

Mining Journal (London).

Mining Magazine (London).

Roskill Information Services Ltd. (London).

Strontium—Supply, Demand, and Technology, U.S. Bureau of Mines, Information Circular 9213, 1989.

Strontium. Ch. in *Mineral Facts and Problems*, U.S. Bureau of Mines, Bulletin 675, 1985.

World Mining, monthly.

---

<sup>1</sup>Prior to January 1996, published by the U.S. Bureau of Mines.

TABLE 1  
SALIENT STRONTIUM STATISTICS 1/

(Metric tons of contained strontium unless otherwise noted 2/)

	1992	1993	1994	1995	1996
<b>United States:</b>					
Production, strontium minerals	--	--	--	--	--
Imports for consumption: 3/					
Strontium minerals	19,700	11,600	16,000	12,700	11,600
Strontium compounds	13,000	15,300	20,000	20,800	20,500
Exports (compounds) 3/	650	260	1,120	1,160	712
Shipments from Government stockpile excesses	--	--	--	--	--
Price, average value of mineral imports at port of exportation, dollars per ton	\$68	\$73	\$68	\$71	\$67
World production 4/ (celestite)	195,000 r/	201,000 r/	233,000 r/	291,000 r/	297,000 e/

e/ Estimated. r/ Revised.

1/ Data are rounded to three significant digits.

2/ The strontium content of celestite is 43.88%, which was used to convert units of celestite.

3/ Source: Bureau of the Census.

4/ Excludes China and the former U.S.S.R.

TABLE 2  
U.S. ESTIMATED DISTRIBUTION OF PRIMARY  
STRONTIUM COMPOUNDS, BY END USE

(Percent)

End use	1995	1996
Electrolytic production of zinc	2	2
Ferrite ceramic magnets	8	9
Pigments and fillers	4	3
Pyrotechnics and signals	7	9
Television picture tubes	75	73
Other	4	4
Total	100	100

TABLE 3  
U.S. EXPORTS OF STRONTIUM COMPOUNDS, BY COUNTRY 1/

	1995		1996	
	Quantity (metric tons)	Value 2/ (thousands)	Quantity (metric tons)	Value 2/ (thousands)
<b>Strontium carbonate precipitated:</b>				
Canada	338	\$313	339	\$304
Japan	51	53	(3/)	6
Malaysia	1,090	362	--	--
Mexico	--	--	24	23
Singapore	--	--	12	11
Other	1	13	6	25
Total	1,480	740	381	368
<b>Strontium oxide, hydroxide, and peroxide:</b>				
Australia	--	--	255	140
Belgium	56	52	19	17
Canada	22	12	21	12
Germany	76	42	84	46
Israel	--	--	26	15
Japan	50	42	165	149
Norway	--	--	25	14
Taiwan	--	--	88	49
Turkey	25	14	--	--
United Kingdom	--	--	491	270
Venezuela	11	14	--	--
Other	9	13	--	--
Total	249	189	1,170	711

1/ Data are rounded to three significant digits; may not add to totals shown.

2/ Customs value.

3/ Less than 1/2 unit.

Source: Bureau of the Census.

TABLE 4  
U.S. IMPORTS FOR CONSUMPTION OF STRONTIUM COMPOUNDS, BY COUNTRY 1/

	1995		1996	
	Quantity (metric tons)	Value 2/ (thousands)	Quantity (metric tons)	Value 2/ (thousands)
<b>Strontium carbonate:</b>				
China	147	\$75	377	\$215
Germany	4,540	2,930	3,360	2,230
Mexico	30,200	18,100	30,500	18,400
Other	18	74	10	104
Total	34,900	21,200	34,300	20,900
<b>Strontium nitrate:</b>				
China	--	--	20	96
France	--	--	5	35
Mexico	160	137	111	138
United Kingdom	11	19	--	--
Other	1	6	--	--
Total	171	162	136	269

1/ Data rounded to three significant digits; may not add to totals shown.

2/ Customs value.

3/ Less than 1/2 unit.

Source: Bureau of the Census.

TABLE 5  
CELESTITE: WORLD PRODUCTION, BY COUNTRY 1/ 2/

(Metric tons)

Country 3/	1992	1993	1994	1995	1996 e/
Algeria e/	5,400	5,400	5,400	5,400	5,400
Argentina	1,200 e/	4,806 r/	8,484 r/	9,325 r/	9,000
Iran e/ 4/	13,138 5/	20,000	20,000	20,000	20,000
Mexico	61,097	71,900	70,000 e/	138,342 r/	143,892 5/
Pakistan	1,448	1,684	2,320 r/	1,625 r/	2,500
Spain	72,295 r/	52,968 r/	102,046 r/	90,972 r/	91,000
Turkey	37,940	43,700	25,000 r/	25,000 e/	25,000
United Kingdom e/	2,000	1,000	--	--	--
Total	195,000 r/	201,000 r/	233,000 r/	291,000 r/	297,000

e/ Estimated. r/ Revised.

1/ World totals and estimated data are rounded to three significant digits; may not add to totals shown.

2/ Table includes data available through July 15, 1997.

3/ In addition to the countries listed, China and the former U.S.S.R. produce strontium minerals, but output is not reported quantitatively, and available information is inadequate to make reliable estimates of output levels.

4/ Data are for year beginning Mar. 21 of that stated.

5/ Reported figure.